THE STRUCTURE OF NEW SESQUITERPENES FROM BASIDIOMYCETES

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Fomannosin<sup>1)</sup>  $\mathcal{I}$  is a metabolite of the wood-rotting fungus, <u>Fomes annosus</u> Karst, which causes decay of the living wood. This substance was reported to have a phytotoxic activity toward seedlings of <u>Pinus tadea</u>.<sup>2)</sup> In the course of an investigation of biosynthesis of the fungal isoprenoids produced by Basidiomycetes, we have isolated two new furanoid sesquiterpenes along with fomannosin  $\mathcal{I}$ , from <u>Fomitopsis insularis</u> which was nonpathogenic and commonly found on pine tree. We assign structure 1 and 3 to these substances from their physicochemical properties.

The compound 1,  $C_{15}H_{20}O_2$  (M<sup>+</sup>, 232) had  $[\alpha J_D + 69.5^{\circ}, \lambda \max 215 nm$  (£, 5100): Ymax 3610, 3490, 876 cm<sup>-1</sup>. The nmr spectrum of 1 shows the signals at  $\delta$  0.88, 1.13, 1.72 (three methyl groups), 4.23 (1H, d, J=11 Hz, a methine proton on a carbon bearing a hydroxyl group), 3.33, 2.88 (ABq, J=16 Hz, an allylic methylene) and 7.04, 7.72. The presence of the  $\beta,\beta$ -disubstituted furan ring was indicated by nmr, ir (876 cm<sup>-1</sup>) uv and mass fragmentation (<u>m/e</u> 81).

Double resonance experiments on the acetate 2 indicate that the following proton arrangement must be involved in 2, coupling constants are shown as follows respectively: Jab=1.8 Hz, Jbc=1.0 Hz, Jcd=10.5 Hz, Jde=7.5 Jdf=9 Hz, Jef 13 Hz, Jga=1.0 Hz, Jah  $\simeq$  0, and Jgh=16 Hz.

$$- \underbrace{\operatorname{Hg}}_{C} \underbrace{\operatorname{Hh}}_{C} \xrightarrow{\operatorname{Ha}}_{C} \underbrace{\operatorname{Ha}}_{C} \xrightarrow{\operatorname{Hb}}_{C} \underbrace{\operatorname{Hc}}_{C} \underbrace{\operatorname{Hd}}_{C} \underbrace{\operatorname{He}}_{C} \underbrace{\operatorname{Hf}}_{C} \underbrace{\operatorname{Hd}}_{C} \underbrace{\operatorname{Hf}}_{C} \underbrace{\operatorname{Hd}}_{C} \underbrace{\operatorname{He}}_{C} \underbrace{\operatorname{Hf}}_{C} \underbrace{\operatorname{Hd}}_{C} \underbrace{\operatorname{He}}_{C} \underbrace{\operatorname{Hf}}_{C} \underbrace{\operatorname{Hd}}_{C} \underbrace{\operatorname{Hf}}_{C} \underbrace{\operatorname{Hd}}_{C} \underbrace{\operatorname{Hf}}_{C} \underbrace{\operatorname{Hd}}_{C} \underbrace{\operatorname{Hf}}_{C} \underbrace{\operatorname{Hd}}_{C} \underbrace{\operatorname{Hf}}_{C} \underbrace{\operatorname{Hd}}_{C} \underbrace{\operatorname{Hd}}_{C} \underbrace{\operatorname{Hf}}_{C} \underbrace{\operatorname{Hd}}_{C} \underbrace$$

The compound 3,  $C_{15H22O3}$  (M<sup>+</sup>, 250), nmr (S) 1.02 (s, 6H), 1.18 (s, 3H), 2.87, 2.65 (ABq, J=16 Hz), 4.60 (d, J=5 Hz), 5.25 (br.s, 2H), 7.10, 7.22 (1H each) gives a crystalline monoacetate 4,  $C_{17H24O4}$  (M<sup>+</sup>, 292), m.p. 125°, nmr (S) 1.06, 1.08, 1.19 (each 3H), 2.08 (s, 3H) 2.46, 2.81 (ABq, J=15 Hz), 5.79 (d, J=8 Hz), 7.09, 7.16. The presence of furan ring in 2 was supported by readily formation of the Diels-Alder adduct with diethyl acetylenedicarboxylate to give compound 6. Treatment of the acetate 4 with phosphorus oxychloride in pyridine



afforded a dehydration product, 5,  $C_{17}H_{22}O_3$  (M<sup>+</sup>, 272):  $\lambda$  max 238 nm (E, 6280). In the nmr spectrum of 5, one of the tertiary methyl signal in 4 (1.19) shifts to 1.88 and a new olefinic proton signal appeared at  $\delta$  6.00 (br. s, 1H). These data show that the structure of 3 has the same skeleton as 1,

From the data mentioned above, the two compounds must have structures  $\downarrow$ and  $\supsetneq$  respectively. It is considered that the compounds have close biogenetical relationship to marasmic acid.<sup>3)</sup> Co-occurrence of  $\downarrow$  and fomannosin  $\Huge{l}$  suggests the biogenesis of both compounds.

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